

**Problem 8 Three elements**

Let A, B and C three elements and a, b and c their valence respectively.

Then:

%A in  $A_bB_a$  is 75 %:

$$\frac{b \cdot A}{b \cdot A + a \cdot B} = 0.75 \quad bA = 3aB$$

%B in  $B_cC_b$  is 7.8 %:

$$\frac{c \cdot B}{c \cdot B + b \cdot C} = 0.078 \quad 11.82cB = bC$$

%C in  $A_cC_a$  is?

$$\frac{a \cdot C}{a \cdot C + c \cdot A} = \frac{a \cdot C}{a \cdot C + \frac{b \cdot C}{11.82 B} \cdot A} = \frac{a \cdot C}{a \cdot C + \frac{3a \cdot B \cdot C}{11.82 B}} = \frac{a \cdot C}{1.254 a \cdot C} = 0.7976 = 79.76$$

An element of the three elements is hydrogen for requirements of electroneutrality in the molecules.

Hydrogen is the element with minor percentage. Then B is hydrogen.

Now  $A = 3a$  and  $C = 11.82c$

$a, c = \{1, 2, 3, 4, 5, 6, 7\}$

$A = 3 \cdot 4 = 12$  then A: carbon

$C = 11.82 \cdot 4 = 47.28$  then C: titanium

In conclusion  $A = C$ ,  $B = H$  and  $C = Ti$ .

The three compounds are  $CH_4$ ,  $TiH_4$  and  $TiC$ .

Solution proposed by

Juan Medina - Master student PUCP - Pontifical Catholic University of Peru