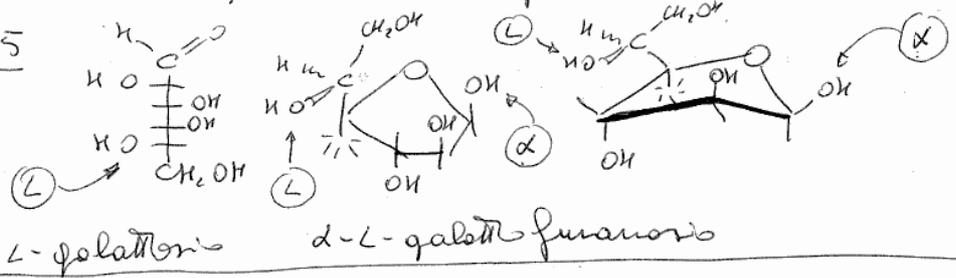
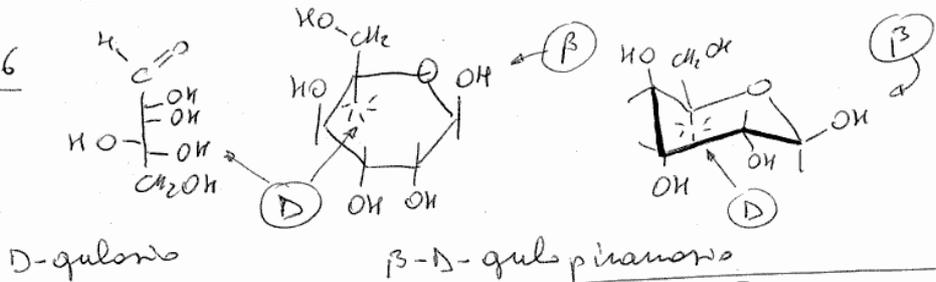


1) scrivi le strutture di Fischer, Haworth e conformazioni:

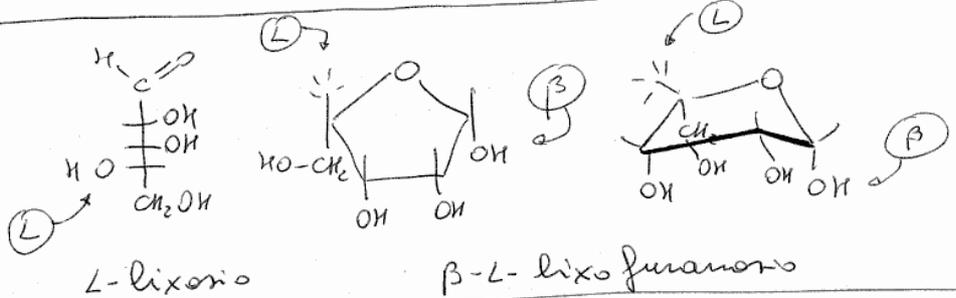
L-galattosio → $\alpha, 5$



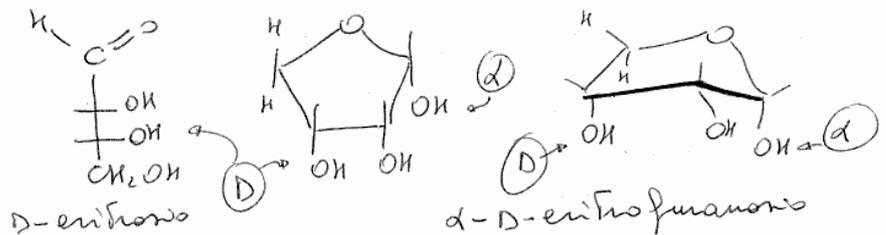
D-galattosio → $\beta, 6$



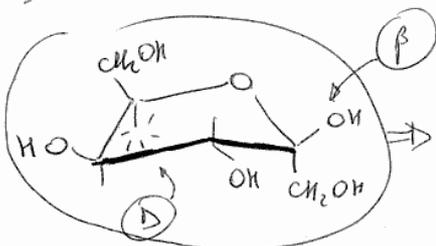
L-lixosio → $\beta, 5$



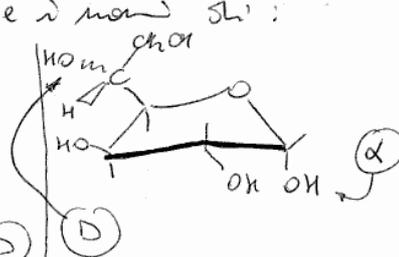
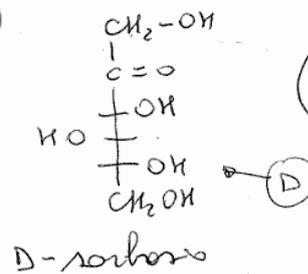
D-eritrosio → $\alpha, 5$



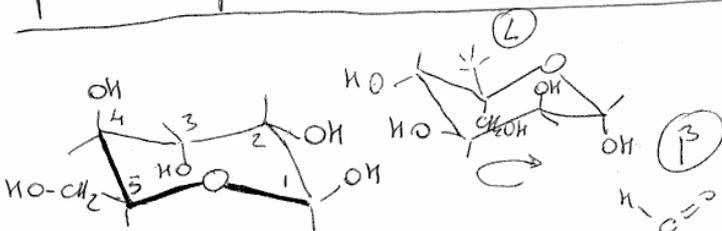
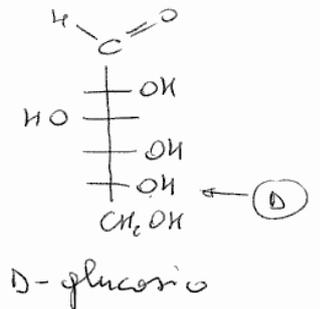
2) Scrivi la struttura di Fischer e i nomi di:



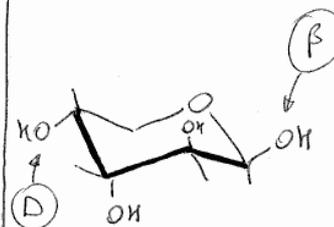
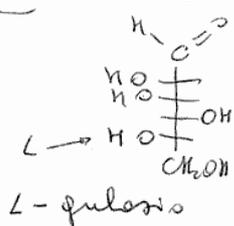
β -D-ribopiranosio



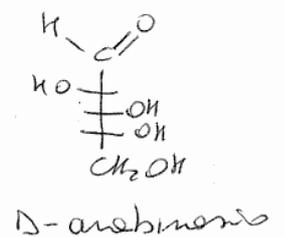
α -D-glucopiranosio

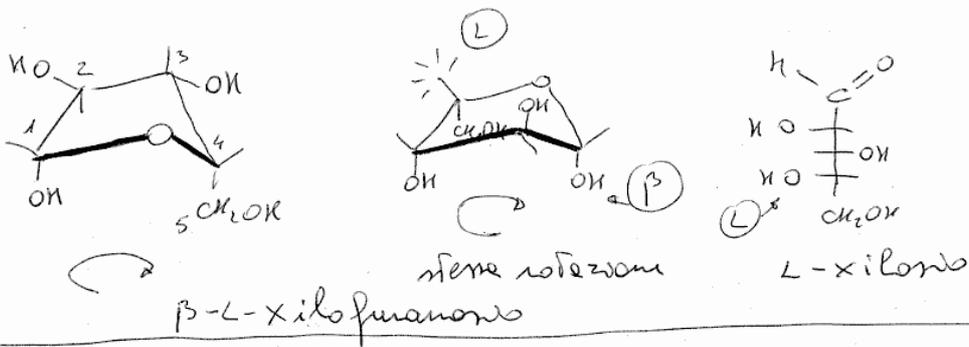


β -L-galattopiranosio



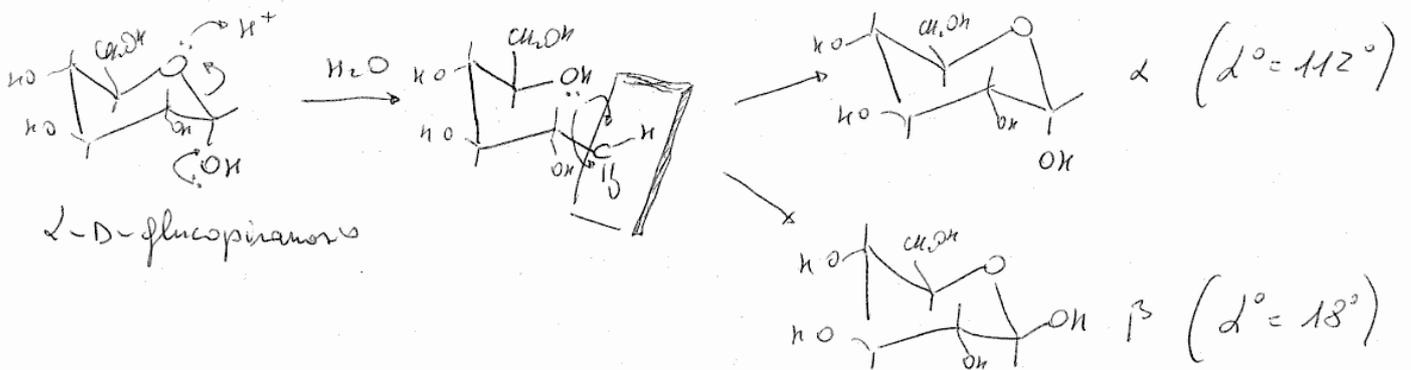
β -D-arabinopiranosio





3) Mutazione: La mutazione è un fenomeno che consiste nel cambiamento di segno e di valore del potere rotatorio di una soluzione di zucchero preparata di fresco. Il potere rotatorio delle soluzioni cambia continuamente fino a portarsi, dopo una o due ore, al valore di equilibrio.

Lo zucchero cristallino è un enomero puro (α o β) e quando viene sciolto in H_2O si permette di determinare il potere rotatorio di quell'enomero puro. Mentre in soluzione si stabilisce l'equilibrio $\alpha \rightleftharpoons \beta$ il potere rotatorio cambia progressivamente fino al valore finale di equilibrio.



4) Se uno zucchero x ha $d^{\circ} = 87^{\circ}$ in l'enomero α e $d^{\circ} = 115^{\circ}$ in β e la soluzione finale ha $d^{\circ} = 91^{\circ}$. Calcolare % di α e β all'equilibrio

$$x \cdot 87 + (1-x) \cdot 115 = 91 \quad 87x + 115 - 115x = 91$$

$$115x - 87x = 115 - 91 \quad 28x = 24 \quad x = 0,857 \quad \alpha = 85,7\%$$

$$\beta = 100 - 85,7 = \beta = 14,3\%$$