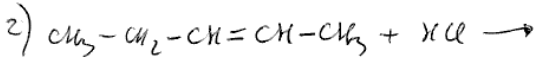
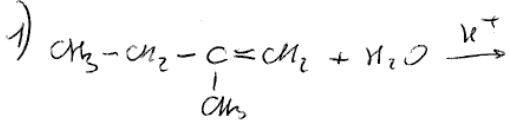
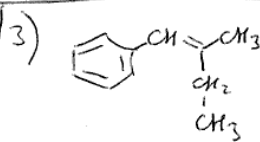


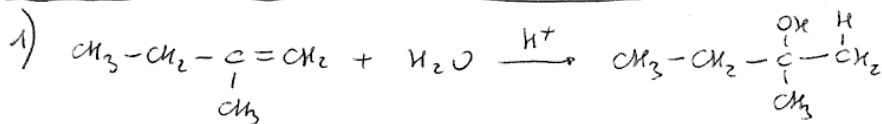
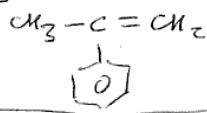
COMPLETE LA REAZIONE:



4) Polimerizzare in ambiente acido:



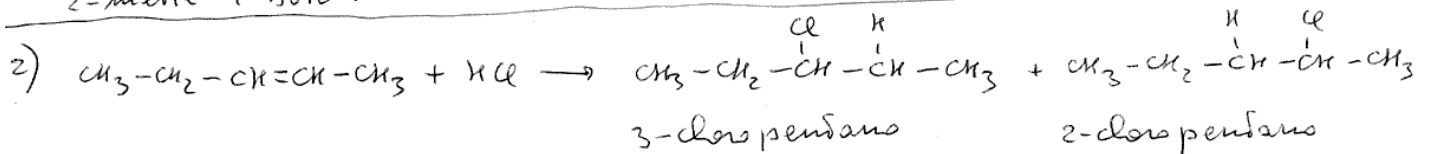
- A) complete la struttura con tutti i C e gli H
- B) scrivi le formule brute contando gli atomi e anche calcolabile
- C) scrivi le reazioni di combustione



2-metil-1-butene

2-metil-2-butanol

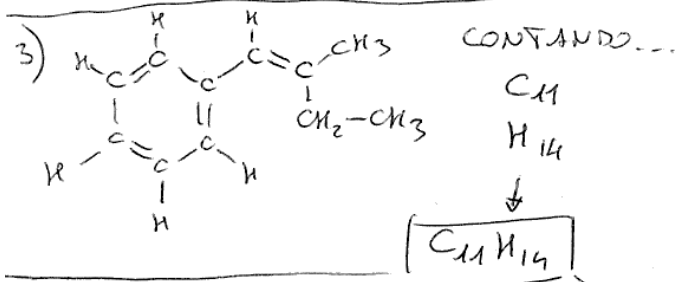
L'OH si lega al C 3° più sostituito (regola di Markovnikov)



3-cloropentano

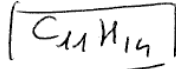
2-cloropentano

in due carboni del doppio legame erano tutti e due reattivi: ottengo 2 prodotti

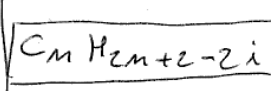


CONTANDO...

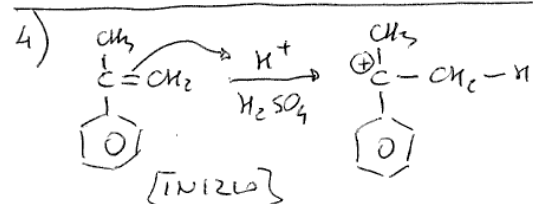
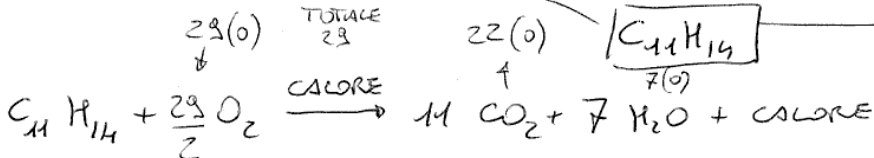
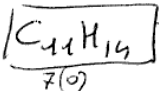
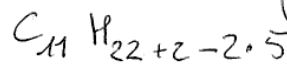
C<sub>11</sub>  
H<sub>14</sub>



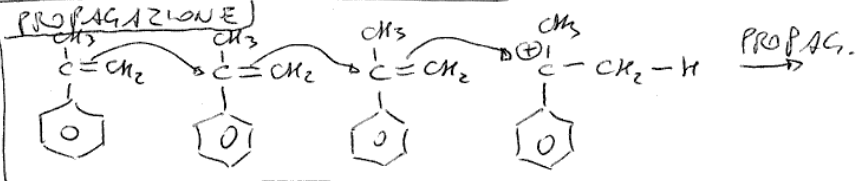
CALCOLANDO...



rispetto all'alcano (C<sub>n</sub>H<sub>2n+2</sub>) si perdono 2 H per ogni doppio legame e per ogni anello (4 doppi legami e 1 anello) 5 INSATURAZIONI

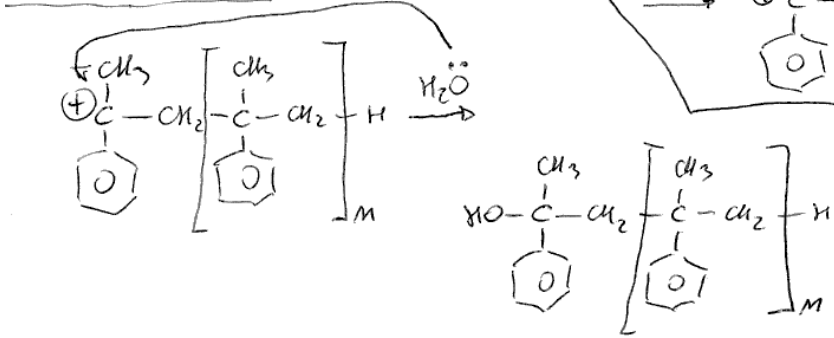


[INIZIO]



PROPAG.

TERMINAZIONE



Polimero finale